

Academic Year	2022/2023
العام الدراسي	
Term	3
الفصل	
Subject	Chemistry /Inspire
العنصر	كيمياء / إنسپر
Grade	10
الصف	
Stream	Advanced
الпотدفق	المتقدم
Number of Main Questions عدد الأسئلة الأساسية	6 Part (1) - 6 Part (2) - 10 Part (3) - 4
Marks per Main Question مarks لكل سؤال اساسي	Part (1) - 5 Part (2) - 5 Part (3) - 5
Number of Bonus Questions عدد الأسئلة الإضافية	2
Marks per Bonus Question مarks لكل سؤال إضافي	5
*** Type of All Questions ملخص نوع كل الأسئلة	Part(1 and 2) MCQ Part (3) FRQ
* Maximum Overall Grade درجة المجموع الكلية	110
Exam Duration - مدة الامتحان	120 minutes
Mode of Implementation - طريقة التقييم	SwiftAssess & Paper-Based
Calculator	Allowed
الإجابة (2)	سموحة

Questions**	Learning Outcome***	Reference(s) in the Student Book (Arabic Version)	
		مثال/تمرين	الصفحة
Part 1 - المحتوى	1 Use the kinetic-molecular theory to explain the behavior of gases	student textbook	240,241,242
	4 List examples for the intramolecular forces	student textbook + Table 2 + figure 9	250,251
	11 Compare the properties of real gases and ideal gases	student textbook + figure 7	293
	14 Compare the properties of suspensions, colloids, and solutions	student textbook	306,307
	15 Describe the electrostatic forces in colloids	student textbook	307,308
	20 Explain the effect of the intermolecular forces on solvation	student textbook + figures 12,13	323,324
	2 Show the effect of mass on the rates of diffusion and effusion	student textbook + Example problem 1+ practice problems	243,244
	3 Calculate the partial pressure of a gas by measuring the total pressure of gases mixture	student textbook + Example problem 2+ practice problems	247,248
	5 Show the effect of intramolecular forces in the properties of matters	student textbook + figure 11 + table 3	253,254
	6 Use the gas laws to solve problems involving the pressure, temperature, and volume of a constant amount of gas	student textbook + figure 1+ example problem 1 + practice problems	278,279,280
	7 Use the gas laws to solve problems involving the pressure, temperature, and volume of a constant amount of gas	student textbook + figure 2 + example problem 2 + practice problems	280,281,282
	9 Relate the amount of gas present to its pressure, temperature, and volume by the ideal gas law	student textbook + Example problem 6 + practice problems	290,291,292
	12 Calculate the amounts of gaseous reactants and products in a chemical reaction	student textbook + example problem 7 + practice problems	296,297
	16 List the types of colloids and solutions	student textbook + figure 3	308
	17 Describe the concentration of solutions by using different units	student textbook + example problem 2 + practice problems	314,315
	18 Describe the concentration of solutions by using different units	student textbook + figure 19 + example problem 5 + practice problems	329,330,331
	8 Relate Avogadro's principle the number of particles of gas to the gas's volume	student textbook + figure 5 + example problem 5 + practice problems	288,289,290
	10 Relate the amount of gas present to its pressure, temperature, and volume by the ideal gas law	student textbook	292,293
	13 Calculate the amounts of gaseous reactants and products in a chemical reaction	student textbook + example problem 8 + practice problems	298,299
	19 Describe the concentration of solutions by using different units	student textbook + figure 8	316
	21 Unpublished	Unpublished	Unpublished
	22 Unpublished	Unpublished	Unpublished
Part 1 (continued) - المحتوى (استكمال)	+ While the overall number of marks is 110, the student's final grade will be out of 100. Example: if a student scores 75 on the exam, the mark will be 75 and if (s)he scores 107, it will be reported as 100 [maximum possible grade].	+ إن جميع الأسئلة 110، فـ درجة الامتحان 100، تـحسب من 100. + على سبيل المثال، إذا حصل الطالب على 75 في الامتحان، فـ درجة الامتحان هي 75، وإذا حصل على 107، فـ درجة الامتحان هي 100 [الدرجة الممكنة]	
	++ Questions might appear in a different order in the actual exam, and bonus questions will be clearly marked on the system (or on the exam paper in the case of G3 and G4).	+ قد تـظهر الأسئلة بـترتيب مختلف في الامتحان الفعلي، وـ سيتم تـحديد الأسئلة الإضافية بشكل واضح على الشـكل (أ) على ورق الامتحان في حالة السـلسـلـات.	
	+++	+ على سبيل المثال، إذا حصل الطالب على 75 في الامتحان، فـ درجة الامتحان هي 75، وإذا حصل على 107، فـ درجة الامتحان هي 100 [الدرجة الممكنة]	
	**** As it appears in the textbook, LMS, and scheme of work (SoW).	+ كما تـظهر في كتاب الطـالـب وـLMS وـمخطط العمل.	

	***** The 2 bonus questions will target LOs from the SoW. These LOs can be within the ones used for the main questions or any other ones listed in the SoW.	+ تستهدف الأسئلة الإضافية 2 الأسئلة التي تـarget المـؤـمـنـات من المـؤـمـنـات. يمكن أن تكون المـؤـمـنـات هذه ضمن المـؤـمـنـات المـؤـمـنـات، أو أي آخـرة مـؤـمـنـات موجودـة في المـؤـمـنـات.	

